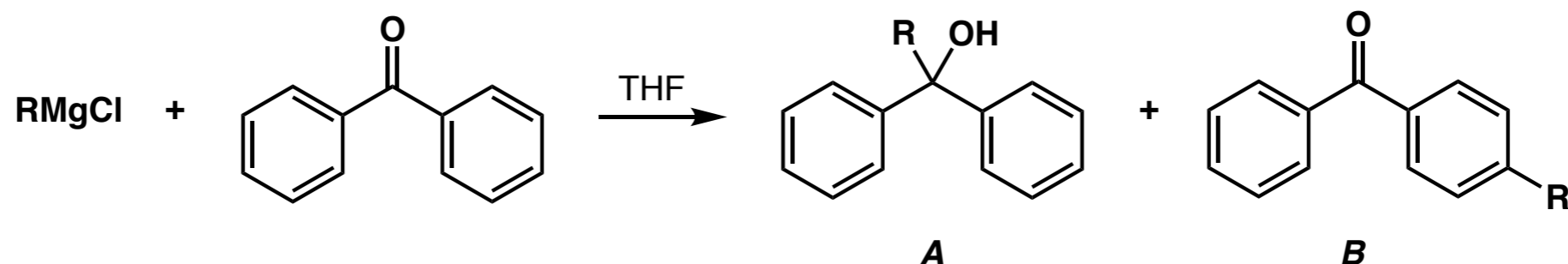


■ It is noticed that,



RMgCl	Yield of A	Yield of B
	100%	0%
	58%	42%

Ashby, E.; Bowers, J. *JACS* **1981**, 103, 2242

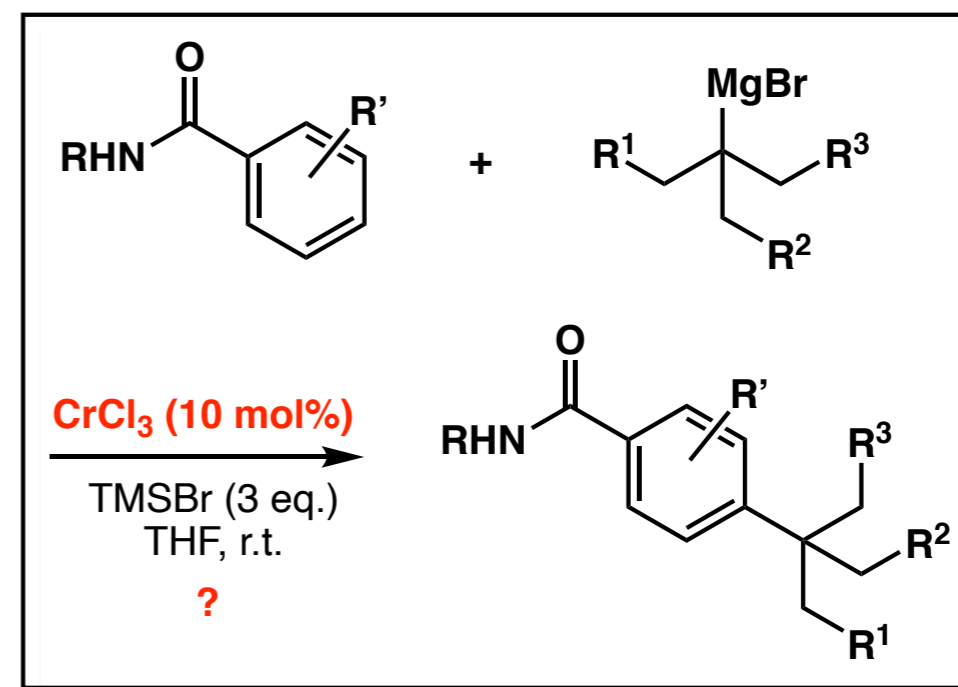
■ Probably because mechanistically,

<i>n</i> -Pr-MgCl	2 e ⁻	2 e ⁻	2 e ⁻
<i>i</i> -Pr-MgCl	—	—	2 e ⁻
<i>t</i> -Bu-MgCl	2 e ⁻	no rxn	SET
Bn-MgCl	2 e ⁻	2 e ⁻	2 e ⁻
Allyl-MgCl	2 e ⁻	2 e ⁻	2 e ⁻
...

R = fluorenylcyclopropyl (radical clock)

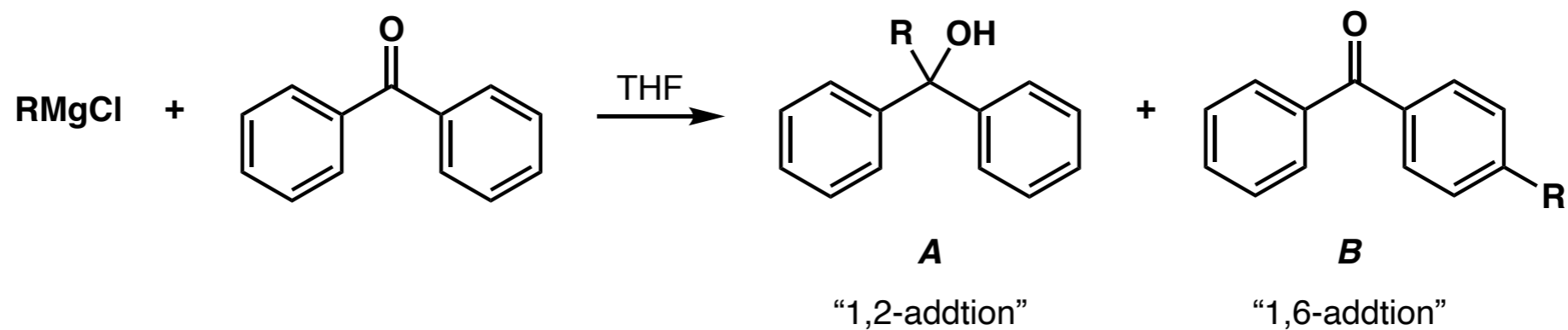
Bartolo, N.; Woerpel, K. *JOC* **2020**, 85, 7848

■ With these mechanistic information, try explaining how CrCl₃(cat.) facilitate this transformation:

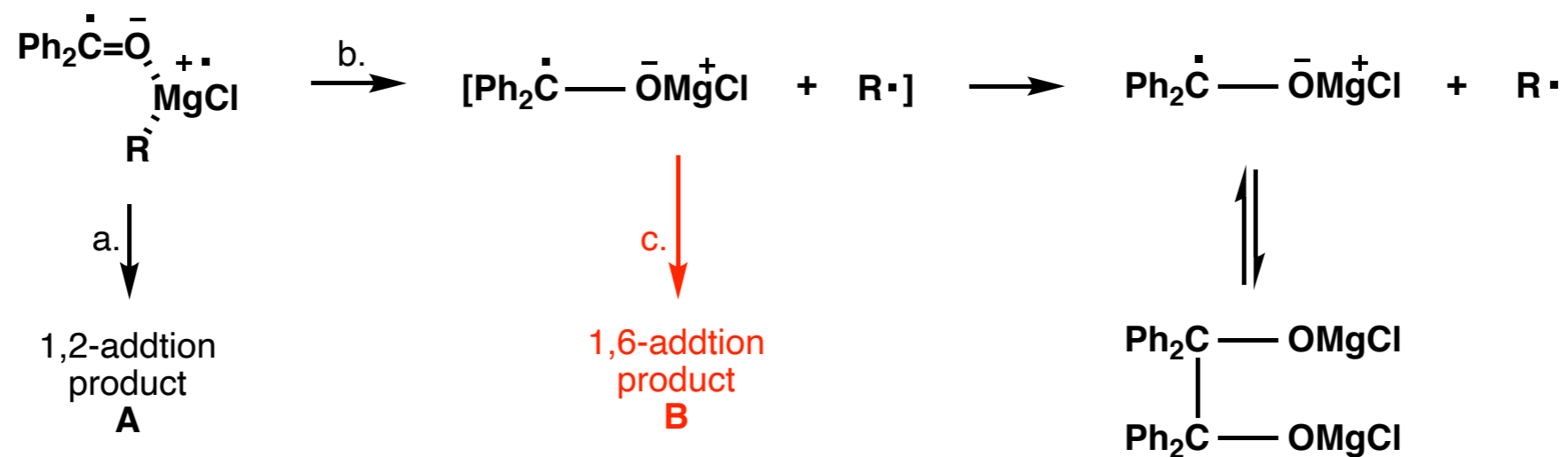
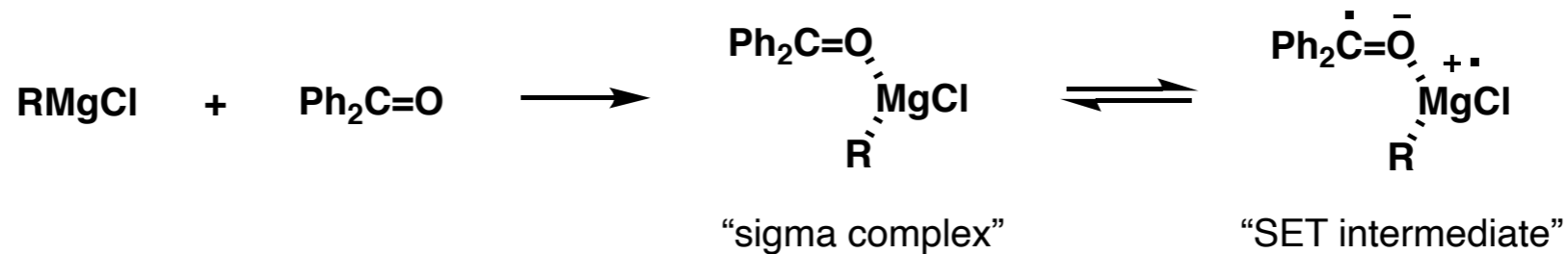


Liu, P., et al. *Nature Comm*, **2018**, 9, 4637

■ Competition between 1,2-addition (path a.) and 1,6-addition products (path c.)

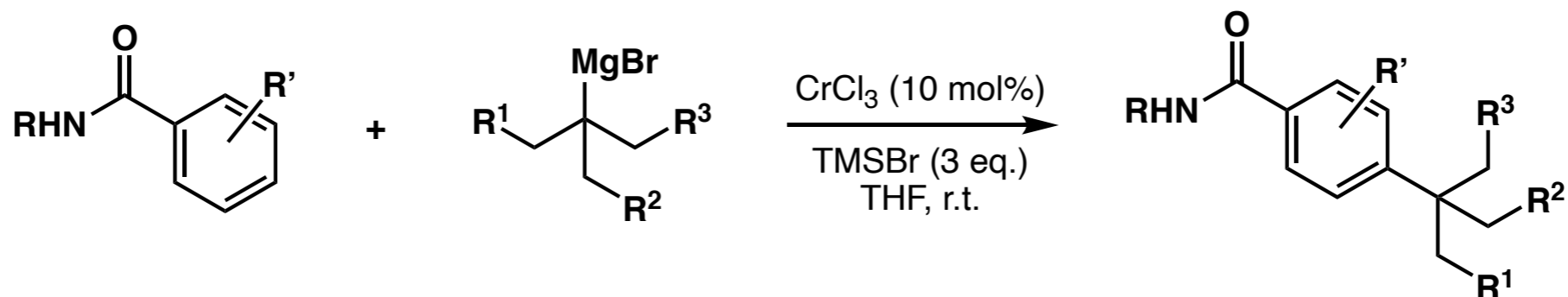


Mechanism:



when R = primary alkyl
rate const. = $10^7 - 10^9 \text{ L mol}^{-1} \text{ s}^{-1}$

■ Enforcement of path c.



Mechanism:

